

1 **Title:** Post-exercise supplementation of sodium bicarbonate improves
2 acid base balance recovery and subsequent high-intensity boxing
3 specific performance
4

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20

21 **Abstract**

22 The aim of this study was to assess the effects of post-exercise sodium bicarbonate (NaHCO₃)
23 ingestion (0.3 g.kg⁻¹ body mass) on the recovery of acid-base balance (pH, HCO₃⁻, and the SID)
24 and subsequent exercise performance in elite boxers. Seven elite male professional boxers
25 performed an initial bout of exhaustive exercise comprising of a boxing specific high-intensity
26 interval running (HIIR) protocol, followed by a high-intensity run to volitional exhaustion
27 (T_{LIM1}). A 75 min passive recovery then ensued, whereby after 10 min recovery, participants
28 ingested either 0.3 g.kg⁻¹ body mass NaHCO₃, or 0.1 g.kg⁻¹ body mass sodium chloride (PLA).
29 Solutions were taste matched and administered double-blind. Participants then completed a
30 boxing specific punch combination protocol, followed by a second high-intensity run to
31 volitional exhaustion (T_{LIM2}). Both initial bouts of T_{LIM1} were well matched between PLA and
32 NaHCO₃ (ICC; $r = 0.94$, $p = 0.002$). The change in performance from T_{LIM1} to T_{LIM2} was greater
33 following NaHCO₃ compared to PLA ($+164 \pm 90$ vs. $+73 \pm 78$ sec; $p = 0.02$, CI = 45.1, 428.8,
34 $g = 1.0$). Following ingestion of NaHCO₃, pH was greater prior to T_{LIM2} by 0.11 ± 0.02 units
35 (1.4%) ($p < 0.001$, CI = 0.09, 0.13, $g = 3.4$), whilst HCO₃⁻ was greater by 8.8 ± 1.5 mmol.l⁻¹
36 (26.3%) compared to PLA ($p < 0.001$, CI = 7.3, 10.2, $g = 5.1$). The current study suggests that
37 these significant increases in acid base balance during post-exercise recovery facilitated the
38 improvement in the subsequent bout of exercise. Future research should continue to explore
39 the role of NaHCO₃ supplementation as a recovery aid in boxing and other combat sports.

40 **Key Words:**

41 **Buffering, alkalosis, acid base balance, combat sports, recovery, nutrition, training**

42 Introduction

43 High levels of glycolytic flux are essential to maintain the required physiological output
44 during combat exercise (Franchini et al., 2011), although a concomitant fall in both muscle and
45 blood pH and bicarbonate ion concentration ($[\text{HCO}_3^-]$) eventually occurs (Fitts, 1994). This is
46 due to the increases in hydrogen ion (H^+) accumulation, which in turn, disturb the state of
47 equilibrium between acidity and alkalinity of body fluids (i.e. acid base balance). Such an
48 alteration is known as metabolic acidosis and has been associated with fatigue by reducing or
49 impairing the release of calcium ions (Ca^{2+}) from the sarcoplasmic reticulum (Messonnier,
50 Kristensen, Juel & Denis, 2007), impeding glycolytic enzyme activity (Atherton, 2003), and
51 altering the strong ion difference leading to reduced action potentials and muscle excitability
52 (Sostaric et al., 2006). The typical daily regimen for a competitive boxer often consists of two
53 sessions comprised of an initial high-intensity intermittent running session followed by a
54 boxing-specific session that mimics the demands of competition, interspersed within a short
55 recovery period (Morton, Robertson, Sutton and Maclaren, 2010). Subsequently a large degree
56 of metabolic acidosis is likely evident in the subsequent bout of exercise, therefore mitigation
57 of the deleterious effects between sessions are prudent to investigate.

58 Pre-exercise ingestion of 0.3 g.kg^{-1} body mass (BM) sodium bicarbonate (NaHCO_3)
59 can lead to an approximate increase in pH ($+0.07 \pm 0.01$) and HCO_3^- from baseline ($+3.9 \pm 0.9$
60 mmol.l^{-1}), eliciting a state of metabolic alkalosis (Carr, Hopkins and Gore, 2011). Ergogenic
61 effects have been reported in combat sports including boxing (Siegler and Hirscher, 2010) and
62 judo (Artioli et al., 2007; Felipe, Lopes-Silva, Bertuzzi, McGinley & Lima-Silva, 2016) by
63 either increasing punches landed or total work done (TWD). Whilst the effects of pre-exercise
64 NaHCO_3 ingestion has been well researched (for review see McNaughton et al., 2016), the
65 effects of post-exercise ingestion between two bouts of exercise to promote recovery has
66 received minimal attention. The use of this alternative method might permit a greater observed
67 improvement in acid base balance during the recovery period, whilst the enhanced level of acid
68 base balance would not have been utilised within the initial bout of exercise. These factors
69 combined might therefore increase performance during the subsequent bout of exercise
70 compared to pre-exercise NaHCO_3 ingestion. Indeed, Gough et al. (2017) reported that 0.3
71 g.kg^{-1} NaHCO_3 ingested 30 mins into a 90 min post-exercise recovery period improved
72 subsequent cycling time to volitional exhaustion by 33 secs ($\sim 14\%$) in recreationally active
73 individuals. It is likely that an enhanced level of acid base balance was the primary mechanism
74 for such an improvement, as the authors reported marked increases in pH and HCO_3^- prior to

75 the second bout of exercise compared to the placebo (pH = +0.07, effect size (ES) = 2.6, HCO_3^-
76 = +7 $\text{mmol}\cdot\text{l}^{-1}$, ES =3.4). It is unknown, however, if these positive findings translate to other
77 exercise modalities such as boxing, and individuals of a higher training status.

78 The mechanisms to explain the performance improvement following NaHCO_3
79 supplementation is not unique to changes in pH and HCO_3^- . Specifically, marked ionic shifts
80 are suggested to contribute to muscle fatigue by impeding maximal Na^+ , K^+ -ATPase activity,
81 subsequently impairing cell membrane excitability (Fitts, 1994; Stephens, McKenna, Canny,
82 Snow & McConell, 2002; Sostaric et al., 2006). Indeed, both large effluxes of extracellular K^+
83 concomitant with reductions in Na^+ have been suggested to exacerbate the K^+ induced decline
84 in force production (Bouclin et al., 1995). Pre-exercise ingestion of NaHCO_3 has been shown
85 to reduce K^+ and increase Na^+ prior to the onset of exercise (Sostaric et al., 2006; Siegler and
86 Hirscher, 2010; Stephens et al., 2002; Jones et al., 2016). Indeed, Siegler and Hirscher (2010)
87 reported NaHCO_3 supplementation prior to a simulated boxing protocol lowered K^+ compared
88 to the placebo condition ($4.0 \pm 0.1 \text{ mEq}\cdot\text{l}^{-1}$ vs. $5.3 \pm 0.4 \text{ mEq}\cdot\text{l}^{-1}$, respectively) and subsequently
89 speculated that this reduction might have facilitated the resulting performance improvement. It
90 is widely argued however, that electrolyte balance should be assessed by the collective analysis
91 of the strong ion difference (SID), which is the balance of the fully dissociated cations and anions
92 in intracellular and extracellular fluid (Stewart, 1983). Synergistic changes in electrolytes are
93 suggested to allow for deeper assessment of fatigue mechanisms, as opposed to reporting
94 changes within a single electrolyte. In the only study to date, Gough et al. (2018a) reported a
95 significant increase in the SID following NaHCO_3 supplementation and an improvement in 2
96 x 4 km time trial cycling bouts interspersed by 40 min recovery, although this study was
97 conducted in a normobaric hypoxic environment. The purpose of this study therefore was to
98 investigate the effects of post-exercise ingestion of NaHCO_3 on acid base balance recovery,
99 the SID and subsequent boxing performance.

100

101 **Materials and methods**

102 Seven male elite professional boxers (age: 27.1 ± 5.1 years, stature: 175.8 ± 5.7 cm,
103 body mass: 72.2 ± 10.3 kg, relative peak oxygen uptake ($\dot{V}\text{O}_{2\text{peak}}$): $55.8 \pm 11.4 \text{ ml}\cdot\text{kg}\cdot\text{min}^{-1}$)
104 from various boxing weight classifications including flyweight, lightweight, junior
105 welterweight (WBO/IBF) super lightweight (WBA/WBC), middleweight & super
106 middleweight completed this study. Participants were considered elite standard boxers and

107 were at least Commonwealth (British Empire), English, International Masters, British Masters,
108 or Midlands Area title holders, with an average of 4.1 ± 3.6 years professional boxing
109 experience. At the time of data collection, all participants were in pre-competition training. The
110 study received institutional ethics committee approval (University of Derby, UK) prior to any
111 testing, and participants were informed of the details of the study, both verbally and in writing,
112 prior to providing written informed consent in accordance with the Declaration of Helsinki.
113 Physical Activity Readiness Questionnaire (PAR-Q) and blood analysis questionnaires were
114 completed prior to each bout of exercise.

115

116 **Preliminary procedures**

117 Prior to each trial, participants were requested to avoid strenuous exercise and to abstain
118 from caffeine and alcohol ingestion for at least 24 hours. Participants were also encouraged to
119 adopt the same mixed balanced diet with adherence monitored through a food diary, which
120 participants recorded 24 hours prior to testing. A photocopy of the food diary was given to each
121 participant to facilitate dietary replication prior to each experimental trial with 100% adherence
122 achieved. Finally, participants were verbally screened to ensure they had refrained from
123 ingestion of ergogenic buffers such as sodium citrate and β -alanine for 6 months prior to
124 beginning the study.

125 Participant's body composition was assessed using Dual Energy X-ray Absorptiometry
126 (Lunar iDXA, GE Healthcare, Hertfordshire, UK) 7-10 days prior to the experimental trials for
127 analysis of body mass (kg). During the same visit, following 3 hours of fasting, participants
128 completed an incremental exercise test on a motorised treadmill (Desmo, Woodway, Germany)
129 to assess peak oxygen uptake ($\dot{V}O_{2\text{peak}}$). Initially participants warmed up for 5 min at $8 \text{ km}\cdot\text{h}^{-1}$
130 with a 0% gradient. The test began with a 3 minute stage at $10 \text{ km}\cdot\text{h}^{-1}$, subsequently the speed
131 increased by $2 \text{ km}\cdot\text{h}^{-1}$ every three min until it reached the $16 \text{ km}\cdot\text{h}^{-1}$ stage. From this point the
132 gradient was increased by 2% every 2 min until volitional exhaustion. Throughout the test
133 expired gas samples were collected via an online breath by breath system (Cortex MetaLyzer
134 II, Biophysik, Leipzig, Germany) which was calibrated before each test as per the
135 manufacturer's guidelines. Expired gas samples were analysed for oxygen consumption ($\dot{V}O_2$),
136 carbon dioxide production ($\dot{V}CO_2$) and respiratory exchange ratio (RER). The highest value of
137 $\dot{V}O_2$ obtained in any 30 second period was used to calculate $\dot{V}O_{2\text{peak}}$.

138

139 **Familiarisation**

140 During the second laboratory visit participants were familiarised with the high intensity
141 interval run (HIIR) protocol (Table 1), and the punch type techniques and combinations (Table
142 2), that would be utilised during experimental trials. In the HIIR, emphasis was placed upon
143 exercising at a percentage of running velocity at $\dot{V}O_{2peak}$ during each differing work interval as
144 opposed to heart rate ensuring the total time at each workload was readily matched. Finally,
145 participants ran at a velocity that elicited 90% $\dot{V}O_{2peak}$ to volitional exhaustion (T_{LIM}) as a
146 measure of high-intensity endurance capacity.

147

148 *** Tables 1 and 2 about here ***

149

150 **Experimental Design and Protocol**

151 Experimental trials were conducted using a repeated measures, partially
152 counterbalanced (due to odd number sample size), double-blind, and placebo controlled design,
153 each separated by seven days. Participants reported for each trial three hours postprandial and
154 at the same time of day to avoid any circadian rhythm effects on performance (Forbes-
155 Robertson, Dudley, Vadgama, Cook, Drawer & Kilduff, 2012). Body mass was measured and
156 recorded at the start of each laboratory visit (Seca 761 weight scales, Birmingham, UK), to
157 monitor possible fluctuations between experimental trials due to the participants being in pre-
158 competition training stages. The following baseline measures were obtained after 5 min seated
159 rest: heart rate (HR; (Polar, FT40, Finland), blood lactate concentration ($[Bl a^-]$), base excess
160 (BE), bicarbonate ion concentration ($[HCO_3^-]$) and a range of electrolytes (sodium $[Na^+]$,
161 potassium $[K^+]$, calcium $[Ca^{2+}]$ and chloride $[Cl^-]$). The electrolyte data was used to calculate
162 the apparent SID using an online spreadsheet (Lloyd, 2004) based on the following formula:
163 $[K^+] + [Na^+] + [Ca^{2+}] + [Na^+] - [Cl^-] - [Bl a^-]$. Blood variables were collected via a finger prick
164 capillary blood sample and analysed with a blood gas analyser (ABL90 Flex, Radiometer, West
165 Sussex, UK). Perceived readiness to exercise (PRE) was then recorded against an 11 point (0-
166 10) scale with 0 representing 'not at all ready to exercise' and 10 representing 'completely
167 ready to exercise' (Higgins et al., 2013).

168 Exercise trials commenced with a 5 minute treadmill run at a velocity eliciting ~60%
169 $\dot{V}O_{2PEAK}$ (warm-up) immediately followed by the HIIR protocol (Table 1) which was repeated
170 three times to imitate the demands of 3x4 minute boxing rounds, each separated by 60 sec

171 active recovery. A self-selected active recovery was recorded and replicated for each recovery
172 interval in both experimental trials. Subsequently, a fourth and final bout was performed on a
173 treadmill at a running velocity eliciting $\sim 90\%$ $\dot{V}O_{2PEAK}$ to volitional exhaustion (T_{LIM1}) with
174 participants blinded from distance and time completed. Overall (i.e. related to cardiovascular
175 strain) ratings of perceived exertion (RPE_O) (Borg scale 6-20; 1982) were recorded within the
176 final 5 sec of each round. Immediately post-exercise HR and RPE_O were recorded. Five min
177 post-exercise HR and blood metabolite/electrolyte data was collected as previously described.

178 Participants then recovered passively for 75 min prior to undertaking subsequent
179 boxing performance. This was selected due to previous data showing this time period is
180 approximately when acid base balance returns to baseline following high-intensity exercise
181 (Gough et al., 2017). Ten minutes into recovery, participants consumed either $0.3 \text{ g}\cdot\text{kg}^{-1}$ body
182 mass of NaHCO_3 or $0.1 \text{ g}\cdot\text{kg}^{-1}$ body mass of sodium chloride (placebo; PLA) within a
183 standardised five minute period. **This time period was selected due to the fear of vomiting if**
184 **ingestion began immediately post-exercise, whilst a longer time period was not used as this**
185 **may have allowed acid base balance to recover back to baseline values prior to ingestion**
186 **(Gough et al., 2017).** Both drinks were mixed in $4 \text{ ml}\cdot\text{kg}^{-1}$ body mass tap water and $1 \text{ ml}\cdot\text{kg}^{-1}$
187 body mass of double strength no added sugar orange squash (Sainsbury's, London, U.K.)
188 (Higgins et al., 2013). Thirty min post exercise abdominal discomfort (AD) and gut fullness
189 (GF) were recorded using an 11 point (0-10) scale, with 0 representing 'empty' and 'completely
190 comfortable', and 10 representing 'bloated' and 'unbearable pain' respectively (Higgins et al.,
191 2013). Water was consumed *ad libitum* during recovery (mean $582 \pm 40 \text{ ml}$).

192 At the end of the 75 minute recovery, HR, PRE, blood metabolites/electrolytes, AD and
193 GF were all recorded prior to participants performing a 5 min standardised dynamic warm up.
194 Participants then completed the boxing specific protocol (Table 2) whereby they were required
195 to strike the focus pads (Serious, Rapid Fire Punch Mitts, London, UK), which were worn by
196 the same researcher for all trials. Each complete cycle consisted of 21 punches with participants
197 instructed to stay in their preferred boxing stance (orthodox or southpaw) throughout. The
198 punch combination cycle was performed repeatedly for 3x3 minute rounds, each separated by
199 60 sec passive recovery. Participants were all given the same boxing gloves (10 oz, Adidas,
200 Hi-Tech Multi-Boxing Glove, Germany) for both experimental trials. An audio and visual
201 boxing gym timer (Title Boxing, De luxe gym timer, USA) kept timing of rounds. Immediately
202 at the end of each round participants HR, RPE_O and ratings of perceived exertion localised to
203 the arms (RPE_A ; Borg scale 6-20) were recorded. Upon completion of the 3 boxing specific

204 rounds AD and GF were also recorded. Following a 60 second rest period, participants then
205 performed a final high intensity treadmill run corresponding to a speed that elicited ~90%
206 $\dot{V}O_{2\text{peak}}$ to volitional exhaustion ($T_{\text{LIM}2}$). Immediately post exercise HR, RPE_O, AD and GF
207 were recorded, and five min post-exercise HR, blood metabolite/electrolytes, AD and GF were
208 recorded.

209

210 **Statistical analysis**

211 Data was firstly checked for normality via a Shapiro-Wilk test, followed by a Mauchly
212 test for homogeneity of variance/sphericity. A paired t test was used for some performance
213 ($T_{\text{LIM}1}$ and $T_{\text{LIM}2}$) and blood/perceptual data (change in HCO_3^- during $T_{\text{LIM}2}$, change in HCO_3^-
214 during recovery, and aggregated GI discomfort). A two-way [treatment \times time] repeated
215 measures ANOVA was conducted with a Bonferroni correction for changes in blood variables
216 (pH, HCO_3^- and lactate). Effect size (ES) for interactions from the ANOVA are reported as
217 partial eta squared ($P\eta^2$), whilst between treatment ES are reported as Hedge's g effect sizes
218 (g) (interpreted as per conventional thresholds described by Cohen 1988). If $p < 0.05$ then 95%
219 CI are reported, where changes that do not cross the zero boundary treated as significant. A
220 Friedman test was used for non-normally distributed data (AD, GF), and where the a priori
221 alpha value was observed (i.e. $p < 0.05$) a post hoc Wilcoxon signed rank-test was conducted
222 with median, z score, and p value reported. For non-normally distributed data the ES is
223 calculated by Z/\sqrt{n} with 0.10, 0.24 and 0.37 considered as small, medium and large effects,
224 respectively (Ivarsson et al., 2013). Reproducibility of the performance in $T_{\text{LIM}1}$ was assessed
225 using intraclass correlation coefficients (ICC), with the r value and significance reported.
226 **Additional statistics such as confidence intervals and effect sizes were used due to the small**
227 **sample size in the study, which might not be suited to statistical procedures such as t test and**
228 **ANOVA in isolation.** Data were analysed using a statistical software package, SPSS (V.24,
229 IBM Inc., Chicago, IL, USA).

230

231 **Results**

232 **Performance**

233 Both initial bouts for $T_{\text{LIM}1}$ were well matched between PLA and NaHCO_3 (328 ± 155
234 vs 307 ± 142 s; ICC: $r = 0.94$, $p = 0.002$; t test, $p = 0.526$), showing that participants were at a
235 similar level of fatigue at the start of the recovery period. Performance in $T_{\text{LIM}2}$ was greater by

236 70 ± 90 sec (28%) following NaHCO₃ compared to PLA (p = 0.084, CI = -153.8, 12.9; Figure
237 1a), with a moderate effect size (g = 0.41). The change in performance from T_{LIM1} to T_{LIM2} was
238 greater following NaHCO₃ compared to PLA (+164 ± 90 vs. +73 ± 78 sec; p = 0.02, CI = 45.1,
239 428.8, g = 1.0; Figure 1b). One participant displayed an ergolytic effect following NaHCO₃
240 ingestion, such that T_{LIM2} decreased by 13% compared to PLA (545 vs. 623 sec). This
241 participant also suffered from moderate to severe GI discomfort.

242

243 **Blood variables**

244 No differences in pH between PLA and NaHCO₃ were observed at baseline (7.43 ±
245 0.04 vs. 7.42 ± 0.02; p = 0.233), or post T_{LIM1} (7.31 ± 0.04 vs. 7.31 ± 0.04; p = 0.696). Following
246 the recovery period, and the ingestion of NaHCO₃, pH was greater prior to T_{LIM2} by 0.11 ± 0.02
247 units (1.4%) (p <0.001, CI = 0.09, 0.13, g = 3.4). Post T_{LIM2}, no difference between treatments
248 was observed for pH (7.31 ± 0.06 vs. 7.33 ± 0.08; p = 0.271; Figure 2a). There were no
249 differences in HCO₃⁻ between PLA and NaHCO₃ at baseline (25.9 ± 1.5 vs. 26.0 ± 1.6 mmol.l⁻¹
250 ¹; p = 0.750), post T_{LIM1} (16.6 ± 2.2 vs. 16.8 ± 2.2 mmol.l⁻¹; p = 0.723), or post T_{LIM2} (17.7 ±
251 3.1 vs. 19.0 ± 3.4 mmol.l⁻¹; p = 0.196). Following recovery however, HCO₃⁻ was greater by 8.8
252 ± 1.5 mmol.l⁻¹ (26.3%) post-NaHCO₃ supplementation compared to PLA (p <0.001, CI = 7.3,
253 10.2, g = 5.1; Figure 2b). The change in HCO₃⁻ during recovery (post T_{LIM1} to pre T_{LIM2}) was
254 greater following NaHCO₃ ingestion compared to PLA (16.6 ± 1.4 vs. 8.0 ± 2.1 mmol.l⁻¹; p
255 <0.001; CI = 6.5, 10.7, g = 4.5). During T_{LIM2}, the change in HCO₃⁻ during exercise was greater
256 for NaHCO₃ compared to PLA (14.3 ± 2.9 vs. 6.9 ± 2.5 mmol.l⁻¹ p <0.001, 10.3, 4.5, g = 2.5).
257 Post T_{LIM2}, BLA⁻ was 5.2 ± 2.6 mmol.l⁻¹ (39.5%) greater following NaHCO₃ (p = 0.002, CI =
258 2.6, 7.3, g = 2.0), with no difference at any other time point (p >0.05; Figure 2c).

259 Ingestion of NaHCO₃ caused marked changes in Na⁺, K⁺, Ca²⁺ and Cl⁻ (Figure 3). A
260 time*treatment interaction was observed for the SID (p = 0.023, Pη² = 0.576), such that a 10%
261 increase in the SID was observed post recovery following NaHCO₃ ingestion compared to PLA
262 (46 ± 1 vs. 36 ± 4 meq/l; p <0.001, CI = 6.3, 13.7, g = 3.2; Figure 4).

263

264 **Heart rate and perceptual measures**

265 Post-exercise ingestion of NaHCO₃ increased HR in rounds 2 and 3 compared to PLA
266 (p <0.05), whilst no effect was observed on RPE_O or RPE_A (p >0.05) during any round of the
267 HIIR. No effect on post T_{LIM2} HR (p = 0.217, g = 0.46) was observed following NaHCO₃.

268 Likewise, no difference in HR between NaHCO₃ and PLA were observed at any time point
269 during T_{LIM2} (all p >0.05). Similarly, NaHCO₃ supplementation had no effect on post T_{LIM2}
270 RPE_O (Z = 1.47, p = 0.383), with no difference observed between treatments at any time point
271 (p >0.05).

272 Abdominal discomfort was greater following NaHCO₃ ingestion at 30 min recovery,
273 displaying a moderate effect size (3.6 ± 3.0 vs. 1.6 ± 2.3; Z = 1.76, p = 0.07, g = 0.7). At the
274 end of recovery, abdominal discomfort had generally reduced, although NaHCO₃ was still
275 greater (1.7 ± 1.7 vs. 0.7 ± 1.3; Z = -1.89, p = 0.06, g = 0.6). No time*treatment interaction was
276 observed for gut fullness (p = 0.219, η² = 0.213). Aggregated GI discomfort was not
277 significantly different between NaHCO₃ ingestion and PLA (19 ± 13 vs. 13 ± 15; p = 0.175, -
278 14.1, 3.2), although it was associated with a moderate effect size (g = 0.40; Figure 5).

279

280 Discussion

281 This study investigated the effects of post-exercise NaHCO₃ ingestion on subsequent
282 high-intensity boxing performance. Following NaHCO₃ ingestion, acid base balance was
283 increased prior to T_{LIM2} compared to PLA which subsequently improved subsequent boxing
284 specific exercise performance. **Athletes and coaches can therefore implement this strategy to**
285 **support training at times when multiple bouts of exercise are carried out with limited recovery**
286 **interspersed.**

287 The findings of the current study show that NaHCO₃ ingestion improved subsequent
288 boxing specific performance, by markedly reducing the decline from T_{LIM1} to T_{LIM2}. This adds
289 to previous work evaluating post-exercise NaHCO₃ supplementation as a recovery supplement
290 (Gough et al., 2017). Indeed, Gough et al. (2017) showed that NaHCO₃ ingestion 30 min into
291 a 90 min recovery period improved subsequent cycling capacity, such that a moderate effect
292 size (g = 0.5) was observed versus the placebo within a group of recreationally trained males.
293 The current study adds however, similar ergogenic effects can be achieved with post-exercise
294 NaHCO₃ ingestion within a shorter recovery time, individuals of a higher training status, and
295 combat exercise. In addition, these findings also support previous literature showing NaHCO₃
296 ingestion is an effective supplement to improve combat performance when ingested prior to
297 exercise (Siegler and Hirscher, 2010; Lopes-Silva et al., 2018). Future research could consider
298 the impact of NaHCO₃ ingestion to enhance subsequent performance in other combat sports.

299 Based on the observed improvements it can be speculated that if NaHCO₃
300 supplementation could be adapted into a chronic weekly supplementation strategy, this might
301 lead to greater adaptation to training. Previous work by Percival et al. (2015) has shown mRNA
302 expression of PGC-1a, a known mechanism for mitochondrial adaptation, was increased 3
303 hours following a high intensity training session with acute NaHCO₃ ingestion compared to a
304 placebo. Based on this evidence it is plausible that this may aid training adaptation in boxing,
305 however, the study by Percival et al. (2015) was in cycling and in lesser trained individuals to
306 the current study (healthy men vs. elite boxers). In addition, other studies investigating the
307 effects of NaHCO₃ ingestion to support training adaptations are equivocal within trained
308 individuals. Indeed, Edge, Bishop and Goodman (2006) reported chronic NaHCO₃ ingestion
309 significantly increased lactate threshold by 11% and time to fatigue (100% VO_{2peak}) by 41%
310 compared to a placebo following 8 weeks of cycling interval training. Both Driller et al. (2013)
311 and Siegler et al. (2018) however, have shown no greater training adaptations following
312 NaHCO₃ ingestion within rowing and resistance exercise modalities across 4 weeks and 10
313 weeks of training, respectively. Considering positive findings have been reported in combat
314 exercise following NaHCO₃ ingestion (Siegler et al., 2010; Lopez-Silva et al., 2018), further
315 research could explore if greater training adaptations occur with chronic NaHCO₃ ingestion.

316 In the present study, the likely mechanism to explain the improvement in subsequent
317 performance is the changes in blood acid base balance between bouts, such that pH, HCO₃⁻ and
318 the SID were significantly higher at 75 min recovery following NaHCO₃ ingestion. Full
319 recovery of pH, HCO₃⁻, and the SID was achieved in approximately 30-35 min. This is in
320 contrast to the placebo condition, which failed to recover any of these blood analytes to baseline
321 within 75 min of recovery. As a result, NaHCO₃ ingestion mitigated the disturbance to acid
322 base balance during T_{LIM2}, which subsequently may explain the performance improvement.
323 Such a greater state of metabolic alkalosis has been shown to increase buffering capacity by
324 facilitating efflux of H⁺ from the active muscle by enhanced circulating HCO₃⁻, and thus,
325 increasing the glycolytic energy contribution to high-intensity exercise (Bishop et al., 2004;
326 Lopes-Silva et al., 2018). The current study supports these mechanisms, reporting a two-fold
327 increase in the HCO₃⁻ change during T_{LIM2}, and a marked increase in lactate post-T_{LIM2}
328 following NaHCO₃ ingestion. Indeed, Lopes-Silva et al. (2018) showed similar changes in
329 post-exercise lactate following NaHCO₃ ingestion, but also reported a significant 31% increase
330 in estimated glycolytic activity during simulated taekwondo combat. It is important to note
331 however, the link between metabolic acidosis and fatigue has been widely criticised, suggesting

332 at physiologically valid muscle temperatures, accumulation of H^+ has limited effects on muscle
333 contractile ability (Westerblad, 2016). As the current study did not assess either temperature or
334 metabolite accumulation in muscle, we cannot confirm that acidosis has a direct impact on
335 fatigue and performance.

336 An alternative mechanism to explain the performance improvement might be the
337 increases in the SID following $NaHCO_3$ ingestion. Reductions in K^+ and Cl^- were observed,
338 whilst Na^+ was increased in the recovery period, which lead to an overall increase in the SID.
339 This could lead to an increase in electrical excitation, membrane potentials and muscle action
340 potentials, which in turn, could support maximal Na^+ , K^+ -ATPase activity (Fitts, 1994; Cairns
341 and Lindinger, 2008). Previous research, however, has suggested the most important electrolyte
342 change is K^+ , by demonstrating that raised extracellular concentration depresses muscle
343 excitability (Cairns and Lindinger, 2008). This suggests that the important changes that
344 $NaHCO_3$ supplementation elicits is in K^+ . Nonetheless, shifts in Cl^- similar to those observed
345 in the present study have been suggested to drive K^+ back to the muscle fibre through inward
346 rectifier channels, which assist in returning the cell back to resting membrane potential
347 (Lindinger and Heigenhauser, 1991). A well-designed study by Bouclin et al. (1995) also
348 showed that when an increased K^+ and reduced Na^+ were altered in combination, the effects on
349 twitch and tetanic contractions were greater than the changes in these ions in isolation. It is
350 more likely therefore, that collective changes in electrolyte regulation explain the ergogenic
351 mechanism of $NaHCO_3$ supplementation. Further research should therefore continue to explore
352 the effects of $NaHCO_3$ supplementation on the SID and exercise performance.

353 One individual presented moderate to high GI discomfort following $NaHCO_3$ ingestion
354 and displayed an ergolytic effect on performance. These findings agree with prior
355 investigations suggesting GI discomfort might be a factor that negates the performance
356 improvement from $NaHCO_3$ (Saunders et al., 2014; Cameron et al., 2010; Deb et al., 2018).
357 Indeed, Saunders et al. (2014) reported upon removing participants who suffered GI discomfort
358 following $NaHCO_3$ ingestion, only then did total work done (TWD) improve ($p = 0.01$, $d =$
359 0.25) compared to when all participants were included ($p = 0.16$, $d = 0.14$). However,
360 performance benefits in combination with the onset of GI discomfort have occurred previously,
361 whilst there is a lack of a direct link between GI discomfort and exercise performance following
362 $NaHCO_3$ ingestion (Higgins et al., 2013; Gough et al., 2018b). Individuals that suffer from
363 severe GI discomfort could benefit from a lower dose of $NaHCO_3$, as $0.2 \text{ g.kg}^{-1} \text{ BM } NaHCO_3$
364 has been shown to produce similar ergogenic responses whilst significantly reducing GI

365 discomfort (Gough et al., 2018b). Alternatively, the athlete could consider gastric bypass
366 methods of delivery (i.e. enteric coated capsules), as novel data has suggested this may be
367 suitable to reduce GI discomfort but still achieve the required increase in acid base balance
368 (Oliveira et al., 2018; Hilton et al., 2019); although the performance responses are currently
369 unclear. Further research should explore both lower doses of NaHCO₃ and the use of gastric
370 bypass methods of delivery to understand the link between GI discomfort and performance
371 following NaHCO₃ ingestion.

372 A limitation of this study is the small sample size, meaning further work is required to
373 establish the impact of manipulating post-exercise acid base balance on performance and
374 recovery. Despite this, the participant cohort were of an elite standard which are typically
375 difficult to access. The current study findings therefore still have high practical application in
376 sports performance, although further research with larger sample sizes are required. These
377 findings compliment previous research investigating NaHCO₃ supplementation and exercise
378 performance within lesser-trained combat athletes (Artioli et al., 2007; Tobias et al., 2013;
379 Lopes-Silva et al., 2017) and support the use of NaHCO₃ supplementation to promote superior
380 recovery.

381 **Conclusion**

382 The use of NaHCO₃ is a suitable ergogenic aid to achieve a greater magnitude of acid
383 base balance recovery and improve subsequent boxing performance within elite level boxers.
384 Being the first study to assess this within an elite participant cohort, the results of this study are
385 of significance to athletes and coaches in an applied setting. **Boxers within the elite category**
386 **could therefore implement this strategy to augment training performance and potentially the**
387 **subsequent adaptations. One participant did present ergolytic effects following NaHCO₃**
388 **ingestion however, which seemed to be due to high GI discomfort. Athletes should therefore**
389 **trial NaHCO₃ ingestion to assess individual tolerability.** Future research should implement
390 similar recovery interventions within a larger sample of elite athletes to explore the
391 effectiveness of NaHCO₃ supplementation as a recovery strategy.

392

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395

396 **Author contributions statement**

397 This study was conceived by MFH and designed by MFH and SR; data were collected by SR
398 and analysed by LAG and MFH; data interpretation and manuscript preparation were
399 undertaken by LAG, MFH, LRM, and AS. All authors approved the final version of the paper.

400

401 **Conflict of interest statement and funding disclosure**

402 None of the authors have any financial interest or benefit arising from the direct applications
403 of this research and there is no conflict of interest. No funding was provided for this study.

404

405 **List of figures**

406 **Figure 1** Overview of performance responses following NaHCO₃ or PLA. * = NaHCO₃ greater
407 than PLA (p <0.05). A = changes between T_{LIM1} and T_{LIM2}, B = Change in performance from
408 T_{LIM1} and T_{LIM2} following NaHCO₃ or Placebo.

409 **Figure 2** Blood acid base balance responses following NaHCO₃ or PLA, where A = pH, B =
410 blood bicarbonate [HCO₃⁻], and C = blood lactate [BLa⁻]. * = NaHCO₃ greater than PLA (p
411 <0.05).

412 **Figure 3** Changes in extracellular electrolytes following NaHCO₃ or PLA, where A = sodium
413 [Na⁺], B = potassium [K⁺], C = calcium [Ca²⁺], and D = chloride [Cl⁻].

414 **Figure 4** Changes in blood strong ion difference (SID) following NaHCO₃ or PLA. * =
415 NaHCO₃ greater than PLA (p <0.05).

416 **Figure 5** Gastrointestinal (GI) discomfort (gut fullness and abdominal discomfort) following
417 NaHCO₃ or PLA. * = NaHCO₃ greater than PLA (p <0.05).

418

419 **List of tables**

420 **Table 1** Example of one round of the high intensity interval run (HIIR) protocol. Key: AR* =
421 Active recovery; SS** = self-selected during familiarisation

422 **Table 2** Punch combinations sequence utilised during boxing specific performance. Key: MIR:
423 move in range, MOR: move out of range, J: Jab, BH: backhand, LU: lead uppercut, BU:
424 backhand uppercut, LH: lead hook, BHH: Backhand hook

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545 **Table 1**

Exercise duration (sec)	~%$\dot{V}O_2$PEAK	Intensity level
30	90	High
30	75	Moderate
30	90	High
30	75	Moderate
30	90	High
30	75	Moderate
30	90	High
30	75	Moderate
60 AR*	SS**	Low

Key: AR* = Active recovery; SS** = self-selected during familiarisation

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557 **Table 2**

Phase 1	Phase 2 (combinations)	Phase 3
MIR	J-	MOR
MIR	J-BH	MOR
MIR	J-BH-LU	MOR
MIR	J-BH-LU-BU	MOR
MIR	J-BH-LU-BU-LH	MOR
MIR	J-BH-LU-BU-LH- BHH	MOR

Key: MIR: move in range, MOR: move out of range, J: Jab, BH: backhand, LU: lead uppercut, BU: backhand uppercut, LH: lead hook, BHH: Backhand hook

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